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A Review on Biological Synthesis of Nanoparticles: Medical and Agricultural Applications

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Abstract: Nanoparticles (NPs) are materials made up of a collection of atoms that have one or more sizes that fall within the range of 1-100 nm on the nanometer scale. Metallic bionanoparticles are currently a very important and dynamic area of research, with significant implications for practical research. Due to its eco-friendly, non-toxic, low energy requirement, and multifunctionality, biological synthesis of NPs is preferable to physical and chemical synthesis. Because they contain significant secondary metabolites that speed up the reduction and stability of the NPs, plants and microbes are effective reducing agents. The created NPs can subsequently be characterization of physical and chemical properties under an electronic microscope (TEM, SEM, AFM) and using spectroscopy (UV-Visible, XRD, IR, etc). A metallic ion or its oxide can be biologically reduced to a nanoparticle quickly, easily, and up to constant temperature and pressure. One of the main causes of illness and mortality in humans is the rise in multi-drug resistance microorganisms brought on by the overuse of antibiotics in individuals who are not diseased. It is crucial that novel antibiotics with various modes of action are developed today to combat bacteria. At low concentrations, metals and its oxides are extremely hazardous to microorganisms. Additionally, widespread illnesses in human and plants are causing grave concern throughout the planet. Due to their wide range of bioactivity, NPs make excellent antibacterial mediators in both the agricultural and medicinal industries. The current research summarizes the creation of NPs via microbial and botanical sources, which allows metals to remain in a stable form and after ionization. Additionally, it displayed an important database on the use of synthetic nanopesticides in many agricultural and medicinal fields.

Keywords: Nanoparticles, Nanopesticides, Agriculture, Antimicrobial Activity, Green synthesis.

1. Introduction :

Nanotechnology is a small area of research, and the word "nano" is derived from the Greek phrase for "dwarf." One billionth of a part is referred to as a "nano"10⁹. Nanotechnology is essentially the application of nanometer-sized particles (1-100 nanometers) [1]. For upcoming applications, nanotechnology has created a variety of cutting-edge materials over the last ten years. One of the biggest issues facing civilization is environmental cleanup, which can be tackled by using nanoparticles [2].

Over the past ten years, there has been a significant increase in the manufacturing of nanoparticles, and now these materials are widely employed in applications such as devices, biomedical, drugs, skincare, power, the environment, catalysis, and materials. significance and opportunity Nanotechnology is one of most lucrative and rapidly developing fields of technology. Recently, the cofriendly nanoparticles synthesis has been considered as

an effective and optimum strategy for further exploitation of bio-organisms as nano factories as the 21st century approaches and billion dollars sales of nano items are expected [3,4].

By employing nanotechnology fertilizers or by promoting plant development, nanomaterials can improve agricultural and medical operations including insecticidal, soil quality, and improve the quality of products. Inside Furthermore, chemicals and carriers based on nanoparticles cut back on the use of pesticides and fertilizers without reducing yield[5].

Metals are reduced to their corresponding oxides by phytochemicals like flavonoids, saponins, quinones, alkaloids, glycosides, polyphenols, amides, tannins, and terpenoids. As decreasing capping, and stabilizing agents, organic substances derived from plants and plant components are used. The biological foundation of green synthesis techniques depends on several factors, including

the solvent, temperature, pressure, and pH levels (acidic, basic, or neutral).

In the straightforward one-step synthesis of NPs, extract and metallic solution are added, and within minutes, nanostructures are formed of the oxidation / reduction processes. [6,7]

On the other hand, nanomaterials are in good quality for a variety of uses, from biomedical to bioenergy. This is because the nano shape is supported by a large surface area that can assist loading of the target molecule for a variety of scientific applications, such as drug carrier for pathogen infections and molecular disorders specially cancer. Analyzed the toxicity of the drug carrier is important by using nanomaterials as drug carriers; this idea led to the development of synthesis and characterization, that can replace chemical processes that result in hazardous nanostructures. Drug distribution can be targeted to the desired region by synthesizing a metal oxide which reacts to a variety of effects [8]. These nanoparticles could be effectively employed in environmental remediation in addition to drug delivery applications since they can decompose pollutants without harming the environment because the nanocarriers are made from natural substances [8]. This research primarily focuses on novel and cutting-edge green synthesis techniques for nanomaterial that are sensitive to a variety of stimuli. These techniques produce prominent and cost-effective nanoparticles which can be used for a variety of applications and have the added benefit of being bio recyclable. This environmentally friendly synthesis not only creates extremely effective nanostructures but also completes its intended task without endangering living things or the environment [9,10].

The current article demonstrates traditional synthetic techniques while emphasizing more recent advancements in environmentally friendly ways to produce metallic nanoparticles, and other significant NPs. The formation methods and the environmental factors that affect the surface shape, size distribution, and other characteristics of these biologically synthesized NPs are then discussed.

2. The Search Methodology

These objectives are three objectives are based on the

huge potential of plants as sources for the synthesis of nanoparticles. it is first necessary to describe the investigation of plants, microorganisms, and their extracts. In the second section, information is given regarding the numerous characterization tools that are employed to locate synthesized NPs. The biological and agricultural uses of the green synthetic NPs are finally explained. Search methodologies and selection criteria for the literature pertaining to scientific journals were meticulously employed. Hindawi, PubMed and others served as comprehensive resources for this review. Research publications spanning the last 15 years were thoroughly consulted.

3. Background history

3.1 Classes of nanoparticles

Chemically, the chemical structure of manufactured nanoparticles, quantum dots, metal-containing nanoparticles (including metal oxides), carbon-based nanoparticles, zero-valent metals, and dendrimers can be divided into five main classes. In recent years, toxicological research has mainly concentrated on the impacts of three of the five kinds of nanoparticles based on their composition: carbon-based nanoparticles (such as carbon nanotubes and fullerenes) and metal or metal-oxide nanoparticles (Ag, Cu, TiO₂ NPs)[9].

3.2 Synthesis of NPs

3.2.1 Views of Nanoparticle Synthesis

As shown in Figure 1, the breakdown (top-down) and build-up (bottom-up) methods have been used to create ultrafine NPs since ancient times.

The breakdown approach is frequently employed during the physiochemical synthesis of NPs. By using physical forces like grinding, pulverization, and other processes, substance is size decreased as a step to the nano size in the degradation process, also known as the thermomechanical technique [10]. Physical processes are hard to access with NPs; normally, microparticles less than 3 m are not difficult to obtain. The second technique for getting NPs is the build-up procedure.

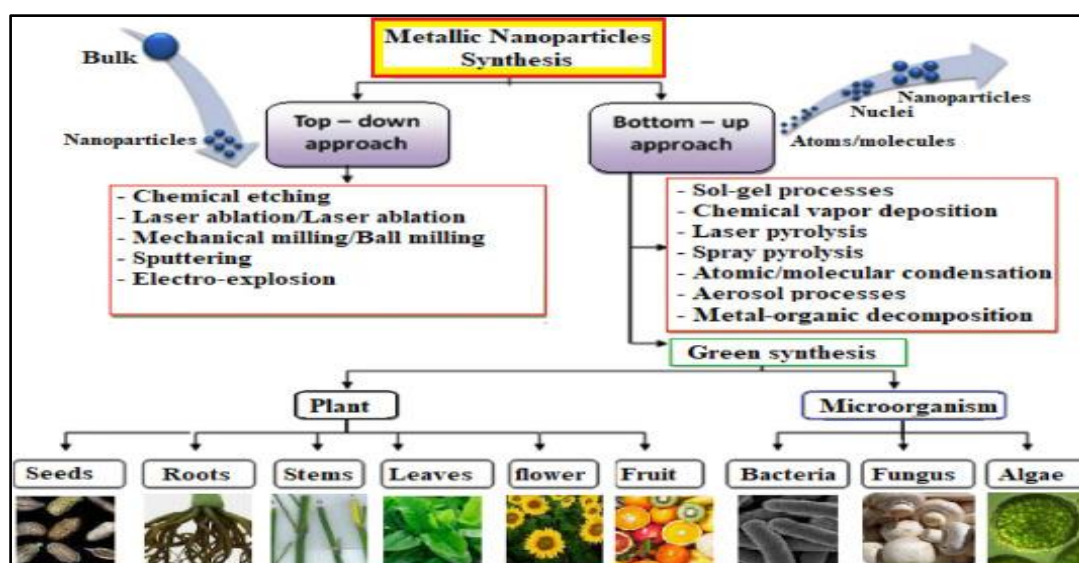


Figure 1. Schematic diagram of Various approaches for the synthesis of NPs.
(This figure has been adapted from ref [11])

Main strategies for the production of Nanoparticles are possible in the liquid and solid phases of matter, without the employment of potentially harmful substances in biogenic synthesis, and with a markedly increased level of chemical synthesis. In the bottom-up method, the applications of the biological process or its components can be observed in the synthesis of biogenic NP. One must take into account a number of characteristics, such as biosynthetic processes, phytochemical content,

biochemical processes, growth factors circumstances, and ideal reaction conditions, in order to select the right organisms or extracts[12].

3.2.2 Secondary Biomolecules for Capping and Stabilization

As shown in Figure2, plant extract serves as a covering and binding agent in addition to being a reductant.

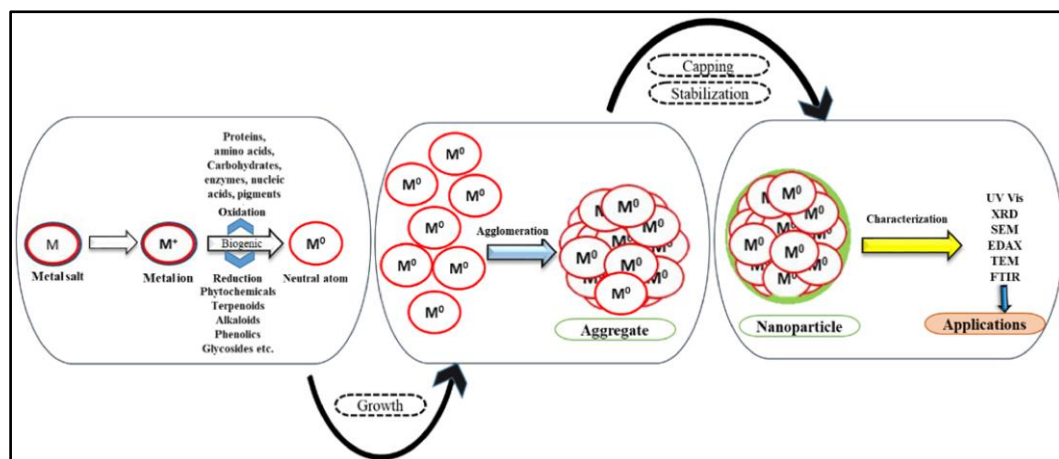


Figure 2. Schematic diagram of biological reduction of NPs. (This figure has been adapted from ref[13]).

When combined with the Fourier Transform Infrared (FTIR) analysis, the Infrared (IR) spectrum of *Camellia sinensis* crude revealed the presence of polyphenols, carboxylic acids, polysaccharides, amino acids, and proteins. This proved the prophetic of biomolecules functioning as covering and stabilizing agents to be correct. [14]. ZnO Nanoparticles displayed peaks in the range of 682-457 cm^{-1} , indicating a higher concentration of phenolics. The stability investigations of silver nanoparticles (Ag NPs) made from *Ziziphora* spp extract at room temperature showed that several metabolite functional groups, including aldehyde, ketone, amine, alcohol, and carboxylic acid compounds, were responsible for the bio nanofabrication of Ag NPs. Significant differences could be seen in the FTIR peak graph between the treatment and untreated samples, which suggested that proteins with amide groups might serve as the covering

layer for metal NPs. [15]. Ascorbic acid's FTIR peak extends inside this OH, CH, C=C circle, and CH_2 wagging revealed that ascorbic acid is present in the *Hibiscus* cannabis extract and is responsible for lowering Ag NPs. The phenolic components in coconut shell extract were produced the synthesis of Au NPs. A big shrub called *Calotropis gigantea* contains phytoconstituents like trisaccharide, glycosides, saponins, beta-sitosterol, alkaloids, flavanols and tannins. The associations of the macromolecules with the Au NPs are indicated by FTIR spectra [16].

3.3 Characterization of Nanoparticles

An essential step in improving nanoparticle synthesis is nanoparticle characterization. The effectiveness of an activity is enhanced by the use of suitable nanoparticles. Various diagnostic methods have been used to study the produced.

Table 1. Illustrated various methods used in the characteristic Nanoparticles

	Instruments	Number	Size Distribution	Agglomeration State	Shape	Surface Area	Chemical Composition
Microscopy techniques	(Scanning Electron Microscopy) SEM	✓	✓	✓	✓		
	(Transmission Electron Microscopy) TEM	✓	✓	✓	✓		
	(Scanning Probe Microscopy) SPM	✓	✓	✓	✓		
Spectroscopy techniques	Visual ultraviolet spectroscopy		✓				✓
	X-ray Diffraction		✓				✓
	FTIR analysis						✓
	Atomic absorption spectroscopy						✓
	Photoelectron X-ray						✓
	Dynamic light Scattering		✓	✓			
	Zeta potential					✓	

NPs based on different features such special functional classes, light scattering and absorption potentials, and so on. In most cases, UV-visible spectroscopy is utilized to confirm the stability and synthesis of NPs.

The features of NPs, including composition and concentration, surface characteristics, surface functional groups, and atomic organization, are measured using Fourier - transform infrared (FTIR) spectroscopy[17]. The location, size, and morphology of NPs can be seen by transmission electron (TEM), scanning electron microscopy (SEM), and atomic force microscope (AFM) [18]. The crystalline structure is determined via X-ray diffraction (XRD). Energy - dispersive x-ray spectroscopy (EDS) is typically apply to determine the elemental analysis of NPs, while the dynamic light scatter (DLS) approach is mostly utilized to show the size and external charge of NPs [19]. All information about the techniques, Instruments, Number,

Size Distribution, Agglomeration, State Shape, Surface Area, and Chemical Composition in Table 1.

3.4 Employment of green synthetic nanoparticles

Green synthesis offers an eco-friendly approach to producing nanoparticles (NPs) by utilizing biological sources such as plants, algae, fungi, and bacteria, thus avoiding toxic chemicals and reducing energy consumption. This method leverages natural biomolecules for reducing and stabilizing agents. Plant-based synthesis (Table 2) uses extracts from various plant parts, while algae-based synthesis (Table 3) employs marine and freshwater algae. Fungi-based synthesis (Table 4) utilizes filamentous fungi and yeast, and bacteria-based synthesis (Table 5) involves both Gram-positive and Gram-negative bacteria, including extremophiles, all contributing to diverse NP applications from biomedicine to environmental remediation.

Table 2. A number of plants used in the synthesis of NPs and their effects

Botanical Names	NPs	image	NP size (nm)	Application	Refs
Ricinus communis	gold		40–80	Antimicrobial Anticancer	20
Anacardium occidentale	gold		10–60	Antimicrobial Anticancer	21
Dracocephalum kotschyi	gold		5–21	Anticancer	22
Euphorbia peplus Fruit	gold		50	• Anticancer • Antibacterial • Insecticidal	23
Gnidia glauca L. Flower	gold		5-20	• Chemocatalytic	24
Abelmoschus esculentus (L.) Moench	gold		45–75	• Antifungal	12
Ananas comosus (L.) Fruit	gold		10 ± 5	• Antimicrobial	25
Cocos nucifera Linn. Liquid endosperm	gold		.022	• Cytotoxicity	26

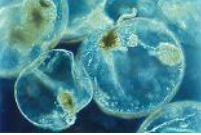
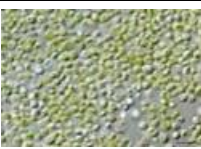
Ficus religiosa L. Bark	gold		20-30	• MTT assay	27
Morinda citrifolia L. Root	gold		12.17– 38.26	• NM	28
Pistia stratiotes L. Aerial and submerged parts	gold		2-40	• NM	29
Senna siamea (Lam.) Leaf	gold		70	• Antibacterial	30
Plumeria alba Linn. Flower	gold		20-30	• Antimicrobial	31
Euphorbia fischeriana Root	gold			• Antioxidant	32
Tropaeolum majus L Flower	silver		38.26	• Antibacterial	33
Camellia sinensis	silver		4-50	• Antibacterial	34
Ferula gumosa, Ferula latisecta, Leaf	silver		20-80	• Antibacterial	35
Teucrium polium, Trachomitum Venetum Leaf	silver		20-80	• Antibacterial	36
Juniperus procera Fruit	silver		30-90	• Antimicrobial	37
Holoptelea integrifolia Unripe fruits	silver		32-38	• Anti-diabetic • Antibacterial • Antiinflammatory	38

Prosopis farcta Leaf	silver		11-15	• Antioxidant • antibacterial	39
Piper longum L. Leaf	silver		28.8	• Antioxidant • Anti-cancer • Antilarvicidal	40
Solidago canadensis Latex	silver		100-300	• Cytotoxic	41
Cocos nucifera Linn. Coir	silver		23 ± 2	• Larvicidal	42
Coriandrum sativum L. Seed	silver		13.9	•Antimicrobial	43
Euphorbia hirta L. Leaf	silver		30-60	• Pesticidal	44
Musa acuminata colla L. flower	silver		Nanoclusters	• Antibacterial • Antifungal	45
Nelumbo nucifera Gaertn. Root	silver		~16.7	• Protein binding • Antioxidant • Anticancer	46
Withania somnifera Linn. Leaf	silver		5-30	•Antimicrobial	47
Ocimum sanctum Linn. Leaf	Silver		50	•Antimicrobial	48
Sapium sebiferum Leaves	platinum		2-12	• Antibacterial	49
Moringa oleifera Leaves	platinum		35	• Antibacterial • Antioxidant • Anticancer	50

Diospyros kaki Leaves	platinum		2-12	NM	51
Lantana camara (L.) Leaves	platinum		35	NM	52
Trianthema portulacastru m Extract	Zinc oxide		25-90	• Cytotoxic • Cytotoxic • Antibacterial • Antifungal • Antioxidant	53
Matricaria Chamomilla	Zinc oxide		40-124	• Antibacterial	54
Lycopersicon esculentum					
Punica granatum Extract	Zinc oxide		32.98– 81.84	• Cytotoxic • Antibacterial	55
Rheum turketanicum Extract	Zinc oxide		17-20	• Cytotoxic	56
Tecoma castanifolia Extract	Zinc oxide		70-75	• Antiseptic • Antioxidant • Antitumor	57
Silybum marianum	Zinc oxide		31	• Cytotoxic • Antifungal • Antibacterial	58
Anchusa italic Flower	Zinc oxide		8-14	• Antibacterial • Cytotoxic	59
Aloe vera Leaves	Zinc oxide		8-20	• Antibacterial • Cytotoxic	60
Rosa canina Fruit	Zinc oxide		50-400	• Antibacterial • Antioxidant • Cytotoxic	61
Boswellia ovalifoliata Bark	Zinc oxide		20	• Antimicrobial	62

Emblca officinalis Fruit	Magnesium oxide		27	• Antibacterial	63
Clitoria ternatea Whole plant	Magnesium oxide		50-400	• Antioxidant	15
Ocimum tenuiflorum Extract	Copper oxide		20-30	• Antibacterial	64
Moringa oleifera Extract	Copper oxide		35-95	• Antifungal	65
Eichhornia crassipes Leaves	Copper oxide		28	• Antifungal	66
Gloriosa superba Leaves	Copper oxide		5-10	• Antibacterial	67
Artocarpus heterophyllus Extract	Titanium dioxide		15-20	• Cytotoxic • antibacterial • anticancer	68
Citrus sinensis fruit	Titanium dioxide		20-50	• Antibacterial, Cytotoxic, and Anticancer	69
Musa alinsanaya Fruit	Titanium dioxide		31	• Larvicidal • antibacterial	70
Psidium guajava Leaves	Titanium dioxide		32	• Antibacterial • Antioxidant	71
Vitex negundo Leaves	Titanium dioxide		93	• Antibacterial	72
Medicago sativa leaves	Samarium		10	• Antitumor	73
Medicago sativa Leaves	Neodymium		10	• NM	74

Table 3. various algae used in the synthesis NPs and their applications

Algae	NPs	Image	size (nm)	Applications in agriculture	Refd.
Amphiroa rigida	Ag		25	Antibacterial and anticancer	75
Chara vulgaris	Ag		16.99 ± 0.3	antibacterial	76
Noctiluca Scintillans	Ag		4-5	Anticancer and antibacterial	77
Botryococcus Brauni	Ag		40-90	Catalyst in synthesis of benzimidazoles	78
Portieria Hornemanni	Ag		-	Antibiotics in the treatment of diseases	79
Ulva armoricana	Ag		12.5	Antibacterial	80
Chlorella Vulgaris	Ag		40-90	Catalyst for the synthesis of quinolines	81
Laminaria Japonica	Ag		20-30	Seed treatment, Pharmacy	82
Spyridia Fusiformis	Ag		Spherical, Rounded, Rectangle	Antibacterial	83
Sargassum Wighti	Ag		18.45–41.59	Pharmacological effect	84

Neodesmus pupukensis	Au and Ag		5-34	Antioxidant, antimicrobial	85
Cystoseira baccata	Au		Spherical	Anticancer	86
Gelidiella acerosa	Au		5 -117	antibacterial, antioxidant	87
Sargassum crassifolium	Au		5-300	Biomedical	88
Anabaena cylindric	ZnO		40-60	Antimicrobial	89
Cladophora glomerata	ZnO		14 -37	Biomedical	90
Tetraselmis indica	ZnO		20-40	Fabric, cosmetic, biomedical, food wrapping	91
Chlorella vulgaris	Pd		15	NM	92
Padina boryana	Pd		11	Antibacterial and anticancer	93
Macrocystis pyrifera	CuO		2-50	NM	94
Botryococcus braunii	Cu and Ag		10-70 (Cu) 40-100 Ag	Antimicrobial	95
Chlamydomonas reinhardtii	CdS		Spherical	Photodegradation of biological dyes	96

Table 4. various fungi used in the synthesis NPs and their applications

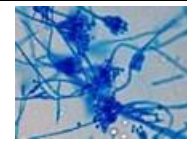
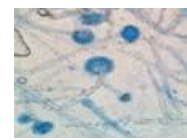
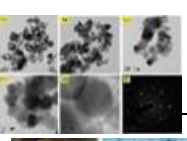
Fungi	NPs	Image	size (nm)	Applications in agriculture	Refd.
Pleurotus sajorcaju	Au,Ag		20-40	NM	97
Phanerochaete chrysosporium	ZnO		50	Antimicrobial	98
Trichoderma Harzianum	Ag		21	Antioxidant, antibacterial	99
Rhizopus oryzae	MgO		20	Germicide, insecticidal, and tanning treatment	100
Morchella esculenta	Au		16	Biomedical	101
Aspergillus sydowii	Ag		1-24	Fungicidal and antiproliferative action to HeLa cells	102
Aspergillus flavus	Cu		2-60	Biomedical	103
Ganoderma lucidum	Ag		15-22	Antimicrobial, antiseptic, antimycotic	104
Periconium sp	ZnO		16-78	Germicide and antioxidant	105
Fusarium solani	Au		40-45	Antitumor	106
Trichoderma harzianum	Ag, CuO		5-18	biotechnological method	107
Fusarium oxysporum	Au		22-30	Therapeutic	108

Table 5. various bacteria used in the synthesis NPs and their application

Bacteria	NPs	NP size (nm)	Application	Refs
<i>Actinobacter spp</i>	Au	5–500	Antimicrobial, fungicidal, nano compost	109
<i>Bacillus subtilis</i>	Au	80	NM	110
<i>Escherichia coli</i>	Au	50	NM	111
<i>L. casei</i>	Au	29	NM	112
<i>Haloferax volcanii</i>	Au	10	Antiseptic activity, Nano biosensors	113
<i>Deinococcus radiodurans</i>	Au	43	Antibacterial action	114
<i>Deinococcus radiodurans</i>	Au	149	Environmental treatment	115
<i>Acintobacter species</i>	Au	15	Antioxidant action	116
<i>Acintobacter species</i>	Au	10	Ecological remediation	117
<i>Plectonema boryanum</i>	Au	10-25	NM	118
<i>Lactobacillus kimchicus</i>	Au	5–30	NM	119
<i>Lactobacillus. casei</i>	Ag	12 –27	Herbal-development stimulator, antimicrobial action, antimycotic effect	120
<i>L. acidophilus</i>	Ag	4–40	NM	121
<i>Pseudomonas stutzeri</i>	Ag	Up to 200	Herbal-development stimulator, antimicrobial action, antimycotic effect	122
<i>Staphylococcus aureus</i>	Ag	160–180	Herbal-development stimulator, antimicrobial action, antimycotic effect	123
<i>Brevibacterium casei</i>	Ag	10–50	NM	124
<i>Escherichia coli</i>	Ag	100	stimulant of plant growth, antibacterial action, and antifungal action	125
<i>Bacillus cereus</i>	Ag	4 - 5	NM	126
<i>Bacillus licheniformis</i>	Ag	18–63	Fungicidal effect	127
<i>Lactobacillus plantarum</i>	Ag	19	NM	128
<i>Lactobacillus rhamnosus</i>	Ag	233	NM	129
<i>Lactobacillus sp.</i>	Ag	15–25	NM	130
<i>Pseudomonas deceptionensis</i>	Ag		antibacterial action and bio film suppression	131
<i>Pseudomonas fluorescens</i>	Ag	5-50	Antibacterial activity and pesticide	132
<i>Sporosarcina koreensis</i>	Ag	varied	Antibacterial action	133
<i>Acinetobacter sp.</i>	Ag	10	Fungicidal - biofilm inhibition	134
<i>Pseudomonas rhodesiae</i>	Ag	20-100	Antibacterial action	135
<i>Pseudomonas poae</i>	Ag	19-44	Antifungal action	136
<i>Streptomyces capillispiralis</i>	Ag	23-63	Antibacterial action	137
<i>Streptomyces zaomyceticus</i>	Ag	11-36	Antifungal action	
<i>Streptomyces pseudogri</i>	Ag	11-44	Larvicidal	
<i>Haloalkaliphilic Streptomyces</i>	Ag	16	Antibacterial action and Antifungal action	138
<i>Streptomyces griseus</i>	Cu	5-50	fungicides	139
<i>Lactobacillus casei</i>	Cu	30–75	Plant fertilizer	140

3.5 Mode of Action of NPs

Various studies indicate that metallic NPs cause damage to microorganisms by generating superoxide radicals (ROS) such hydroxyl and superoxide anion, which destroy cell

walls and membranes as well as cause DNA damage and other oxidative and destructive effects. Figure3 displays a diagram representation of the NPs action process.

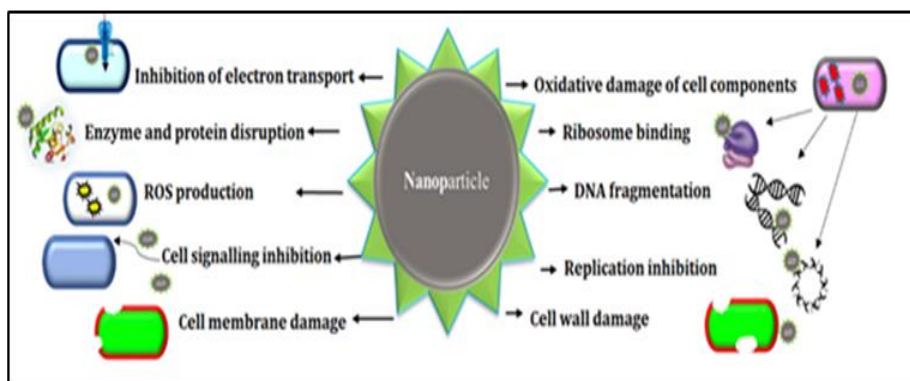


Figure 3. Schematic diagram of mechanism of action of NPs against microorganisms.
(This figure has been adapted from ref. [13])

4. Conclusions and Future Perspectives:

A paper illustrated the most recent advances in bioactive nanoparticle (NP) green production, characterization, and applications. It also explains how secondary metabolites generated from plants and microbes, like alkaloids, terpenoids, quinones, others and, are applied as capping and reducing agents in these environmentally friendly methods. These bioactive compounds are organically formed from safe plant materials including algae, fungi, and bacteria, as well as plant extracts from many widely available species like bananas, tea, onions, and coconuts. The avoidance of hazardous or toxic substances can be overcome by green synthesis; also, synthesis on a reasonable scale can be accomplished. Fine size control, which increases the stability and lifespan of biologically synthesized NPs over chemical and physical alternatives, is their main advantage. Based on the different metal kinds, several methods have been suggested for simpler nanoparticle generation. Due to its less well-known mechanism of action, physiological studies of NPs are significant in vitro experiments, but there are few data on in vivo studies. Due to its quick production in the liquid state and its collaborative action against microorganisms, a biostatic analysis of utilized metals reveals extensive utilization. However, the region-specific, varietal, and part time variations in the chemical components of plant crude extract may frequently result in different outcomes in various laboratories, which calls for additional research into identifying important plant phytochemicals or metabolites in order to develop more modern processes for NP synthesis. Nutritional value, global food production, plant defense, disease detection in plants and animals, and environmental monitoring can all benefit from nanotechnology.

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مراجعة التخليق البيولوجي للجسيمات النانوية: تطبيقات طبية وزراعية

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المخلص: الجسيمات النانوية (NPs) هي مواد تتكون من مجموعة من الذرات التي يتراوح حجم واحد أو أكثر منها بين 1-100 نانومتر على مقياس النانومتر. تُعد الجسيمات النانوية الحيوية المعدنية حاليًا مجالًا بحثيًا مهمًا وديناميكيًا للغاية، ولها تأثيرات كبيرة على البحث العملي. نظرًا لكونه صديقًا للبيئة، وغير سام، ومنخفض متطلبات الطاقة، ومتعدد الوظائف، فإن التخليق البيولوجي للجسيمات النانوية مفضل على التخليق الفيزيائي والكيميائي. تُعد النباتات والميكروبات عوامل اختزال فعالة لأنها تحتوي على مستقلبات ثانوية مهمة تسرع اختزال واستقرار الجسيمات النانوية. يمكن بعد ذلك توصيف الجسيمات النانوية المتكونة بالخصائص الفيزيائية والكيميائية تحت المجهر الإلكتروني (TEM)، SEM، AFM وباستخدام المطيافية (UV-Visible، IR، XRD)، إلخ. يمكن اختزال أيون معدني أو أكسيده بيولوجيًا إلى جسيم نانوي بسرعة وسهولة وعند درجة حرارة وضغط ثابتين. أحد الأسباب الرئيسية للمرض والوفاة لدى البشر هو تزايد الكائنات الدقيقة المقاومة للأدوية المتعددة، والنتائج عن الإفراط في استخدام المضادات الحيوية لدى الأفراد غير المصابين بالمرض. من الأهمية بمكان تطوير مضادات حيوية جديدة ذات آليات عمل مختلفة اليوم لمكافحة البكتيريا. بتركيزات منخفضة، تكون المعادن وأكاسيدها شديدة السمية للكائنات الدقيقة. بالإضافة إلى ذلك، تسبب الأمراض المنتشرة في الإنسان والنبات قلقًا بالغًا في جميع أنحاء الكوكب. نظرًا لنطاقها الواسع من النشاط البيولوجي، تُعد الجسيمات النانوية وسائط ممتازة مضادة للبكتيريا في كل من الصناعات الزراعية والطبية. يلخص البحث الحالي تكوين الجسيمات النانوية عبر المصادر الميكروبية والنباتية، مما يسمح للمعادن بالبقاء في شكل مستقر بعد التأين. بالإضافة إلى ذلك، عرضت الدراسة قاعدة بيانات مهمة حول استخدام المبيدات النانوية الاصطناعية في العديد من المجالات الزراعية والطبية.

الكلمات المفتاحية: الجسيمات النانوية؛ المبيدات النانوية؛ الزراعة؛ النشاط المضاد للميكروبات؛ التخليق الأخضر.